

The First Total Synthesis of Preverecynarmin

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Abstract Preverecynaimin, isolated from a pennatulatean coral, has been synthesized from E,E-farnesol. © 1999 Elsevier Science Ltd. All rights reserved.

Preverecynarmin 1, (+)-(1E,3E,7E,11E)-cembra-1,3,7,11-tetraen-6-yl acetate was first isolated in 1990 from both *Armina maculate* and its prey, the pennatulacean coral *veretillum cynomorium* along with three other briarane diterpenoids and cembrene-C 2 (1E,3E,7E,11E)-cembra-1,3,7,11-tetraene.² It is first time that cembranoids have been isolated from pennatulaceans. The co-occurrence of both the briarane and cembrane skeletons supports the theory that the cembranoid carbon skeleton is a biosynthetic precursor of the briaranes. This prompted our search for a practical synthesis of preverecynarmin. As far as we know, neither biological activity nor a total synthesis of 1 has been reported. Herein we wish to describe the first total synthesis of (\pm) -preverecynarmin (\pm) -1.

Our synthetic route which started from E,E-farnesol involves three key steps: 1) alkylation of the cyanohydrin trimethylsilyl ether 5 with halide 6; 2) the regionselective epoxidation of 9; 3) the intramolecular macrocyclization of 12 induced by Ti(0).

a) MnO₂, *n*-hexane, r.t., 15h., 97%; b) Me₃SiCN, KCN/18-crown-6,0°C, 30min, 100%; c) 1.LiN(SiMe₃)₂, THF, 0°C, 20min then **6**, r.t., 4h, 23%; 2. 10% aq. n-Bu₄N⁺F/THF, r.t., 4h, 95%; d) NaBH₄, MeOH, 89%; e) TBDPSiCl, imidazole, DMF, r.t., 4h, 89%; f) mCPBA(0.6eq), CH₂Cl₂, r.t., 30min, 56%; g) HClO₄, dry CH₂Cl₂, r.t., 5min, 75%; h) 1. *p*-TsOH, MeOH, r.t., 1h, 90%; 2. MnO₂, *n*-hexane, r.t., 8h, 94%; I) Zn/TiCl₄, Py, THF, reflux 24h. 66%; j) 1. 1M *n*-Bu₄N⁺F⁻ in THF, r.t., 20h; 2. AcO₂, Py, DMAP, r.t., 30min, 78%

The first stage of the synthesis is the construction of a 20-membered carbon chain possessing a carbonyl group at C-6. Although many papers have reported that sulfur-stabilized anions can be used as acyl carbonion equivalents in alkylation reactions,³⁻⁶ we selected a cyanohydrin TMS ether as the acyl anion equivalent.⁷ Farnesol was oxidized by MnO₂ in *n*-hexane to farnesal **4**, which was converted to the cyanohydrin trimethylsilyl ether **5** by an addition of Me₃SiCN in the presence of a catalytic amount of KCN/18-crown-6 complex.⁸ The cyanohydrin **5** was treated with 1.25equiv of LiN(SiMe₃)₂ in THF and the lithiated cyanohydrin alkylated with **6**⁹ to afford the alkylated cyanohydrin which was directly converted into the ketone by using a catalytic amount of *n*-Bu₄N⁺F⁻ in 10% aqueous THF without purification. After reduction with NaBH₄, the alcohol **8** so obtained was protected with TBDPSCI to afford the silyl ether **9**.

Another key step was the regioselective epoxidation of the silyl ether **9**. When 0.6eq mCPBA was used, the major product was the epoxide **10**. In order to obtain more of the epoxide **10**, this reaction was repeated three times. In anhydrous CH₂Cl₂, HClO₄ converted the epoxide **10** into the ketone **11**. Removal of the THP group from **11** with a catalytic amount of *p*-TsOH in MeOH followed by oxidation using 20 equiv. of MnO₂ in *n*-hexane resulted in formation of the cyclization precursor **12**. To effect cyclization, a highly diluted solution of **12** in 30 ml DME was syringed slowly over 24hrs to a mixture of TiCl₄/Zn-DME. The macrocyclization product **13** was then deprotected using 1M *n*-Bu₄N⁺F⁻ in THF and then acetylated by Ac₂O in pyridine in the presence of a catalytic amount of DMAP to give the title compound (±)-**1** as clear oil. The spectral data of (±)-**1** agreed with that of literature. The state of the compound (±)-1 as clear oil.

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Reference and Notes

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- 9. The iodide 6 was prepared from 3-methyl-2-buten-1-ol in three steps. First the alcohol was protected with DHP, then oxidised by TBHP in presence of 0.2 eq. SeO₂, and then the allylic alcohol was iodinated with I₂/PPh₃/imidazole.
- 10. a). McMurry, J. E., Chem. Rev., 1989, 89, 1513; b). The cyclization precursor 12 (120mg, 0.215mmol) was dissolved in 30 ml DME. This solution was added to a mixture of TiCl₄/Zn in 20 ml DME. The concentration of substrate is no more than 5×10⁻³mol/l.
- 11. Spectral data of compound (±)-1: v=2959, 2921, 2860, 1736, 1438, 1358, 1240; ¹H NMR: δ (400MHz, C_6D_6)=1.03 (d, 3H, J=7.0Hz, CH₃), 1.04 (d, 3H, J=7.0Hz, CH₃), 1.53 (s, 3H, CH₃), 1.58 (s, 3H, CH₃), 1.78 (s, 3H, CH₃), 2.02 (s, 3H, CH₃CO), 2.10-2.60 (m, 11H, CH₂, CH), 5.05 (m, 1H, CH=), 5.23 (t, 1H, J=9.1Hz, CH=), 5.90 (m, 1H, CHO), 6.16 (m, 2H, CH=); ¹³C NMR: δ (100MHz, C_6D_6)=16.4, 17.3, 18.2, 21.0, 21.7, 23.1, 24.8, 28.0, 32.9, 37.3, 39.5, 46.1, 70.2, 119.1, 124.1, 124.6, 126.3, 130.0, 134.3, 140.0, 147.0, 169.6; m/z (EI, 70ev): 330 (M⁺, 20%), 270 (52), 255 (7), 227 (10), 202 (14), 187 (20), 159 (81), 136 (42), 121 (100);